## (3) If particles are fermions:

Because temporarily I am forgetting about the spin, then I cannot place 2 particles in  $n_1 = 1$ .

Then, the true ground state (E=5K) becomes:

$$\psi(x_1, x_2) = \frac{\sqrt{2}}{a} \left[ \sin(\pi x_1/a) \sin(2\pi x_2/a) - \sin(2\pi x_1/a) \sin(\pi x_2/a) \right]$$

$$\hat{P} \psi(x_1, x_2) = \psi(x_2, x_1) = -\psi(x_1, x_2)$$



x = 0 at left wall of box.



	In summary (if spin is not included):		
	distinguishable	bosons	fermions
E=2 <i>K</i>	1 state	1 state	0 state
E=5 <i>K</i>	2 states	1 state	1 state
etc			

These differences has profound implications for many-body physics (condensed matter, nuclear, ...) and for thermodynamics

## Three particles? (spin still not included)

## distinguishable



 $\infty$ 

x = 0 at left wall of box.



x = 0 at left wall of box.





x = 0 at left wall of box.



x = 0 at left wall of box.

fermions

bosons





x = 0 at left wall of box.



x = 0 at left wall of box.

## 5.1.2 Exchange Forces

This is an "effective force", not a real one.

It is another of the consequences of having symmetric or antisymmetric combinations. Consider 1D and two different states a and b.

distinguishable

$$\psi(x_1, x_2) = \psi_a(x_1)\psi_b(x_2)$$

1

bosons

$$\psi_{+}(x_{1}, x_{2}) = \frac{1}{\sqrt{2}} [\psi_{a}(x_{1})\psi_{b}(x_{2}) + \psi_{b}(x_{1})\psi_{a}(x_{2})]$$

fermions

$$\psi_{-}(x_{1}, x_{2}) = \frac{1}{\sqrt{2}} [\psi_{a}(x_{1})\psi_{b}(x_{2}) - \psi_{b}(x_{1})\psi_{a}(x_{2})]$$

Consider the typical "distance" between particles, via the quantity:

$$\langle (x_1 - x_2)^2 \rangle = \langle x_1^2 \rangle + \langle x_2^2 \rangle - 2 \langle x_1 x_2 \rangle$$
  
Distance in 1D

$$\langle x_1^2 \rangle = \int x_1^2 |\psi_a(x_1)|^2 dx_1 \int |\psi_b(x_2)|^2 dx_2 = \langle x^2 \rangle_a$$

$$\langle x_2^2 \rangle = \int |\psi_a(x_1)|^2 dx_1 \int x_2^2 |\psi_b(x_2)|^2 dx_2 = \langle x^2 \rangle_b$$

$$\langle x_1 x_2 \rangle = \int x_1 |\psi_a(x_1)|^2 dx_1 \int x_2 |\psi_b(x_2)|^2 dx_2 = \langle x \rangle_a \langle x \rangle_b$$

$$\langle (x_1 - x_2)^2 \rangle_d = \langle x^2 \rangle_a + \langle x^2 \rangle_b - 2 \langle x \rangle_a \langle x \rangle_b$$

The main "punchline" is that the distance we just deduced actually **changes** for the symmetric and antisymmetric combinations (see proof on page 204):

$$\begin{array}{l} \left| (x_1 - x_2)^2 \right|_{\pm} = \langle x^2 \rangle_a + \langle x^2 \rangle_b - 2 \langle x \rangle_a \langle x \rangle_b \mp 2 |\langle x \rangle_{ab} |^2 \\ \text{fermions} & \text{fermions} & \text{fermions} \end{array} \right|^2$$

So even non-interacting particles (no Coulomb repulsion at all) suffer an effective force: bosons attract and fermions repel, in the sense that bosons are closer than classical particles and fermions are further apart.

This is a quantum effect, no classical analog!

Sec. 5.1.3. Now all together: the space dependence and the spin (still in 1D for simplicity)

$$\psi_{S}(\mathbf{r}_{1},\mathbf{r}_{2}) = \frac{1}{\sqrt{2}} [\psi_{a}(x_{1})\psi_{b}(x_{2}) + \psi_{b}(x_{1})\psi_{a}(x_{2})]$$
  
$$\psi_{AS}(\mathbf{r}_{1},\mathbf{r}_{2}) = \frac{1}{\sqrt{2}} [\psi_{a}(x_{1})\psi_{b}(x_{2}) - \psi_{b}(x_{1})\psi_{a}(x_{2})]$$

$$\chi_{\mathsf{S}}(\mathsf{S}_{1},\mathsf{S}_{2}) = \frac{1}{\sqrt{2}} \left( \uparrow_{1} \downarrow_{2} + \downarrow_{1} \uparrow_{2} \right) \qquad \chi_{\mathsf{AS}}(\mathsf{S}_{1},\mathsf{S}_{2}) = \frac{1}{\sqrt{2}} \left( \uparrow_{1} \downarrow_{2} - \downarrow_{1} \uparrow_{2} \right)$$

For 2 electrons (i.e. fermions) in two different states:

$$\Psi = \psi_{S}(\mathbf{r}_{1},\mathbf{r}_{2}) \chi_{AS}(\mathbf{S}_{1},\mathbf{S}_{2}) \text{ OK } ! \qquad \Psi = \psi_{S}(\mathbf{r}_{2}, \chi_{S}(\mathbf{S}_{1},\mathbf{S}_{2}) \text{ NO}$$
$$\Psi = \psi_{AS}(\mathbf{r}_{1},\mathbf{r}_{2}) \chi_{S}(\mathbf{S}_{1},\mathbf{S}_{2}) \text{ OK } ! \qquad \Psi = \psi_{AS}(\mathbf{r}_{1},\mathbf{r}_{2}) \chi_{AS}(\mathbf{S}_{1},\mathbf{S}_{2}) \text{ NO}$$

If a=b, then only  $\Psi = \psi_{S}(\mathbf{r}_{1},\mathbf{r}_{2}) \chi_{AS}(\mathbf{S}_{1},\mathbf{S}_{2})$  possible (Pauli !!).

