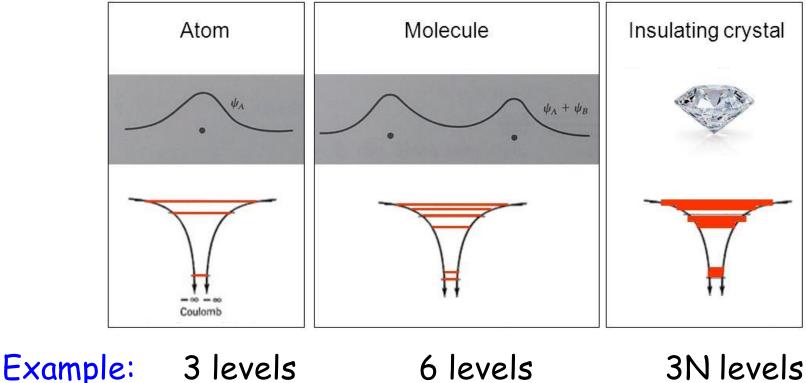


When atoms are brought together, the wave functions start overlapping.

When that happens, energy levels split by small amounts, keeping the total number of levels constant.



"bands" are formed

The inner electrons, like those in (1s), remain trapped near its original nucleus. But the outer electrons, with more extended wave functions, can move from atom to atom.

Depending on how the bands are formed, and how many electrons you have, you may have a metal or insulator.

We will consider two crude approximations:

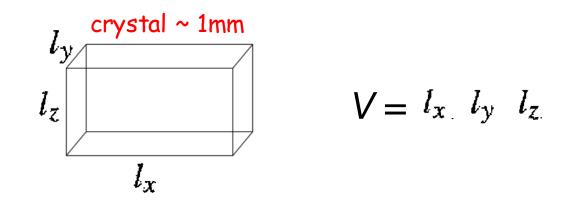
(a) Non-interacting electrons in a giant box (such as a cubic well)

(b)Non-interacting electrons in a periodic potential produced by nuclei.

5.3.1 The Free Electron Gas

Considering the enormity of the problem, we need to start making crude approximations and see how they work, and refine accordingly.

We will first assume that the "conduction electrons" -those that can jump from atom to atom -- are totally free inside a giant box (i.e. a 3D well) the size of the crystal.



 $V(x, y, z) = \begin{cases} 0, & \text{if } 0 < x < l_x, \quad 0 < y < l_y, \text{ and } 0 < z < l_z \\ \infty, & \text{otherwise.} \end{cases}$

This is the plan within this very crude approximation:

(1) We will solve the Sch Eq for ONE PARTICLE, i.e. a particle in a big box. Very easy! It was already done in a HW problem.

$$-\frac{\hbar^2}{2m}\nabla^2\psi = E\psi$$

(2) After solving this problem, we will simply start filling levels with 2 electrons per state (1 up, 1 down), until we have the (large) number of electrons we want. The e-e repulsion will be neglected.

(3) Conceptually, it is the analog of using the exact solution of the H atom, but for a much heavier element like Cu (29) or Au (79).

So let us recall fast the steps to solve the "cubic well":

(a) Use "separation of variables" $\psi(x, y, z) = X(x)Y(y)Z(z)$ plug into Sch Eq, divide by XYZ, and find the equation is written as 3 independent pieces.

$$-\frac{\hbar^2}{2m}\frac{d^2X}{dx^2} = E_x X; \quad -\frac{\hbar^2}{2m}\frac{d^2Y}{dy^2} = E_y Y; \quad -\frac{\hbar^2}{2m}\frac{d^2Z}{dz^2} = E_z Z$$

$$E = E_x + E_y + E_z$$

(b) Introduce standard "k" notation:

$$k_x \equiv \frac{\sqrt{2mE_x}}{\hbar}, \quad k_y \equiv \frac{\sqrt{2mE_y}}{\hbar}, \quad k_z \equiv \frac{\sqrt{2mE_z}}{\hbar}$$

(c) Solutions are:

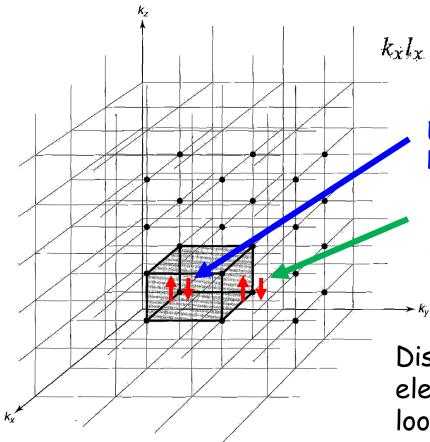
$$X(x) = A_x \sin(k_x x) + B_x \cos(k_x x)$$
$$Y(y) = A_y \sin(k_y y) + B_y \cos(k_y y)$$
$$Z(z) = A_z \sin(k_z z) + B_z \cos(k_z z)$$

(d) Boundary conditions X(0) = Y(0) = Z(0) = 0lead to cancellations

$$B_x = B_y = B_z = 0$$

(e) Other boundary conditions are $X(l_x) = Y(l_y) = Z(l_z) = 0$ lead to $k_x l_x = n_x \pi$, $k_y l_y = n_y \pi$, $k_z l_z = n_z \pi$ $n_x = 1, 2, 3, ..., n_y = 1, 2, 3, ..., n_z = 1, 2, 3, ...$

$$\psi_{n_x n_y n_z} = \sqrt{\frac{8}{l_x l_y l_z}} \sin\left(\frac{n_x \pi}{l_x}x\right) \sin\left(\frac{n_y \pi}{l_y}y\right) \sin\left(\frac{n_z \pi}{l_z}z\right)$$
$$E_{n_x n_y n_z} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_x^2}{l_x^2} + \frac{n_y^2}{l_y^2} + \frac{n_z^2}{l_z^2}\right) = \frac{\hbar^2 k^2}{2m} \quad \mathbf{k} \equiv (k_x, k_y, k_z)$$

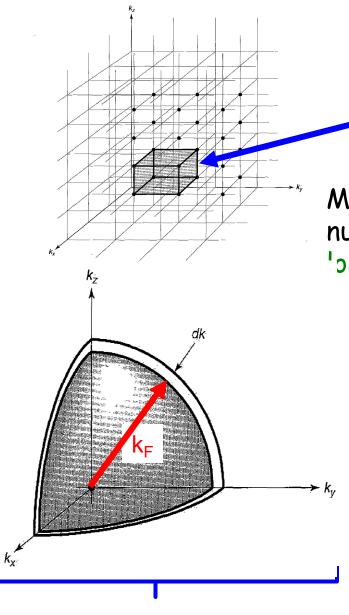


$$k_x l_x = n_x \pi, \quad k_y l_y = n_y \pi, \quad k_z l_z = n_z \pi$$

Each point is an eigenstate. Lowest energy is $(n_x, n_y, n_z)=(1,1,1)$

Next levels are (n_x, n_y, n_z)=(1,2,1) ... deg=3 -> (2,1,1),(1,2,1),(1,1,2)

Discrete but there are so many electrons that from far away it looks like a continuum ...



The ground state is an octant with a high density of points

Each shaded block in this grid occupies $\frac{\pi^3}{l_x l_y l_z} = \frac{\pi^3}{V}$

Most electrons in atoms are trapped near the nucleus, but some (q=1,2,3,...) are free to move 'ong distances: the conduction electrons.

There is a huge number of atoms N even in a small crystal $(1mm)^3$

$$\frac{1}{8} \left(\frac{4}{3} \pi k_F^3 \right) = \frac{Nq}{2} \left(\frac{\pi^3}{V} \right)$$
Octant vol. in
the "continuum" Vol. of a block
Number of
blocks needed

Repeating:
$$\frac{1}{8}\left(\frac{4}{3}\pi k_F^3\right) = \frac{Nq}{2}\left(\frac{\pi^3}{V}\right)$$

Now define the density of electrons:

$$\rho \equiv \frac{Nq}{V}$$

The Fermi wavevector is:

$$k_F = (3\rho\pi^2)^{1/3}$$

The Fermi energy is: This in practice is a key number!

$$E_F = \frac{\hbar^2}{2m} (3\rho\pi^2)^{2/3}$$

The Fermi energy is NOT the ground state energy but the energy of the electrons with the individual highest energy.