5.4.3 The Most Probable Configuration

In thermal equilibrium, we are searching for the most probable configuration $(N_1, N_2, ..., N_n, ...)$ i.e. the one that can be achieved in the largest number of ways i.e. such that $Q(N_1, N_2, ..., N_n, ...)$ is the largest.

The catch is that there are constraints:

$$\sum_{n=1}^{\infty} N_n = N \qquad \sum_{n=1}^{\infty} N_n E_n = E$$

Maximizations with constraints are handled via a method called Lagrange Multipliers:

The general problem is that of maximizing a function of many variables $F(x_1, x_2, x_3, ...)$ subject to constraints such as $f_1(x_1, x_2, x_3, ...) = 0, f_2(x_1, x_2, x_3, ...) = 0$, etc.

The address this problem we introduce a bigger function with Lagrange multipliers $G(x_1, x_2, x_3, \dots, \lambda_1, \lambda_2, \dots) \equiv F + \lambda_1 f_1 + \lambda_2 f_2 + \cdots$

And request:
$$\frac{\partial G}{\partial x_n} = 0$$
 $\frac{\partial G}{\partial \lambda_n} = 0$

Lagrange multipliers

Better to work with InQ instead of Q since they have maxima at same point, but InQ is easier

$$G \equiv \ln(Q) + \alpha \left[N - \sum_{n=1}^{\infty} N_n \right] + \beta \left[E - \sum_{n=1}^{\infty} N_n E_n \right]$$

Use Sterling formula: $\ln(z!) \approx z \ln(z) - z$ for $z \gg 1$

Again, only results will be given here, for the most probable configuration. For distinguishable particles:

$$\ln(Q) = \ln(N!) + \sum_{n=1}^{\infty} \left[N_n \ln(d_n) - \ln(N_n!) \right]$$

$$N_n = d_n e^{-(\alpha + \beta E_n)}$$

For identical fermions:

$$\ln(Q) = \sum_{n=1}^{\infty} \{\ln(d_n!) - \ln(N_n!) - \ln[(d_n - N_n)!]\}$$

$$N_n = \frac{d_n}{e^{(\alpha + \beta E_n)} + 1}$$

For identical bosons:

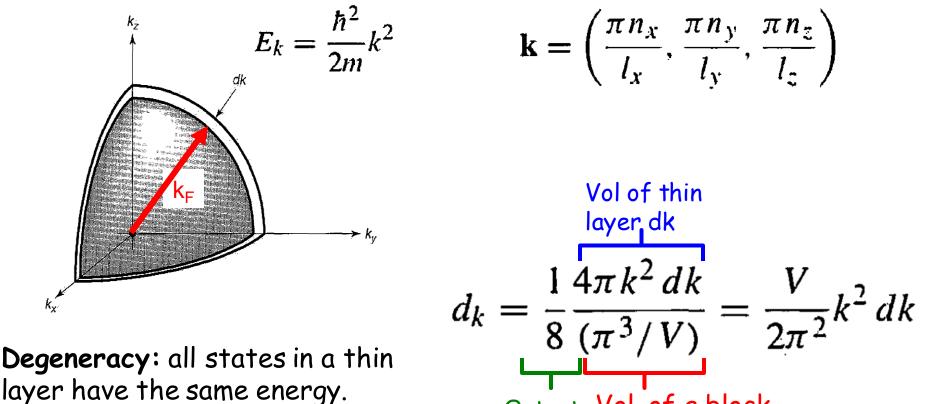
$$\ln(Q) = \sum_{n=1}^{\infty} \left\{ \ln[(N_n + d_n - 1)!] - \ln(N_n!) - \ln[(d_n - 1)!] \right\}$$

$$N_n = \frac{d_n - 1}{e^{(\alpha + \beta E_n)} - 1}$$

<u>5.4.4 Physical Significance of α and β .</u>

Thus far, α and β are merely Lagrange multipliers, but they have an important physical meaning.

To understand the meaning, we will use the example of the ideal gas employing a 3D infinite square well.



layer have the same energy.

The first constraint is:

For distinguishable particles:

$$\sum_{n=1}^{\infty} N_n = N$$

$$N_n = d_n e^{-(\alpha + \beta E_n)}$$

$$E_k = \frac{\hbar^2}{2m} k^2$$

$$d_k = \frac{1}{8} \frac{4\pi k^2 dk}{(\pi^3/V)} = \frac{V}{2\pi^2} k^2 dk$$

$$N = \frac{V}{2\pi^2} e^{-\alpha} \int_0^\infty e^{-\beta \hbar^2 k^2 / 2m} k^2 \, dk = V e^{-\alpha} \left(\frac{m}{2\pi \beta \hbar^2}\right)^{3/2}$$

Rearranging numbers:

$$e^{-\alpha} = \frac{N}{V} \left(\frac{2\pi\beta\hbar^2}{m}\right)^{3/2}$$

The second constraint is:

Again, for distinguishable particles:

$$\sum_{n=1}^{\infty} N_n E_n = E$$

$$E_k = \frac{\hbar^2}{2m} k^2$$

$$N_n = d_n e^{-(\alpha + \beta E_n)} = E_k = \frac{\hbar^2}{2m} k^2$$

$$d_k = \frac{1}{8} \frac{4\pi k^2 dk}{(\pi^3/V)} = \frac{V}{2\pi^2} k^2 dk$$

$$E = \frac{V}{2\pi^2} e^{-\alpha} \frac{\hbar^2}{2m} \int_0^\infty e^{-\beta \hbar^2 k^2/2m} k^4 dk = \frac{3V}{2\beta} e^{-\alpha} \left(\frac{m}{2\pi \beta \hbar^2}\right)^{3/2}$$

$$e^{-\alpha} = \frac{N}{V} \left(\frac{2\pi \beta \hbar^2}{m}\right)^{3/2}$$

$$e^{-\alpha} = \frac{N}{V} \left(\frac{2\pi \beta \hbar^2}{m}\right)^{3/2}$$

$$e^{-\alpha} = \frac{N}{V} \left(\frac{2\pi \beta \hbar^2}{m}\right)^{3/2}$$

$$F = \frac{3N}{2\beta} \quad \text{vs} \quad \frac{E}{N} = \frac{3}{2} k_B T$$

$$\beta = \frac{1}{k_B T}$$

Instead of α , we will introduce a quantity called the chemical potential via a definition:

$$\mu(T) \equiv -\alpha k_B T$$

Thus, $\alpha = -\mu \beta$ which simplifies the math.

$$N_n = d_n e^{-(\alpha + \beta E_n)} \longrightarrow \frac{N_n}{d_n} = e^{-(-\mu\beta + \beta E_n)} = e^{-(\varepsilon - \mu)\beta} = n(\varepsilon)$$

Most probable number of particles in a (one-particle) state with energy ε :

$$n(\epsilon) = \begin{cases} e^{-(\epsilon - \mu)/k_BT} & \text{MAXWELL-BOLTZMANN} \\ \frac{1}{e^{(\epsilon - \mu)/k_BT} + 1} & \text{FERMI-DIRAC} \\ \frac{1}{e^{(\epsilon - \mu)/k_BT} - 1} & \text{BOSE-EINSTEIN} \end{cases}$$

Here, we will only discuss the case of fermions where both the temperature and the chemical potential can be easily visualized. For classical and bosonic particles, T is the same as usual while μ is somewhat related to how many particles you have. μ is crucial in chemical reactions, where number of particles is not conserved.

Consider fermions: as
$$T \rightarrow 0$$

 $e^{(\epsilon - \mu)/k_BT} \rightarrow \begin{cases} 0, & \text{if } \epsilon < \mu(0) \\ \infty, & \text{if } \epsilon > \mu(0) \end{cases}$
 $n(\epsilon) \rightarrow \begin{cases} 1, & \text{if } \epsilon < \mu(0) \\ 0, & \text{if } \epsilon > \mu(0) \end{cases}$
 $n(\epsilon) \rightarrow \begin{cases} 1, & \text{if } \epsilon < \mu(0) \\ 0, & \text{if } \epsilon > \mu(0) \end{cases}$
 $n(\epsilon) \rightarrow \begin{cases} 1, & \text{if } \epsilon < \mu(0) \\ 0, & \text{if } \epsilon > \mu(0) \end{cases}$
 $r(\epsilon) \rightarrow \begin{cases} 1, & \text{if } \epsilon < \mu(0) \\ 0, & \text{if } \epsilon > \mu(0) \end{cases}$
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 $r(\epsilon) \rightarrow \begin{cases} 1, & \text{if } \epsilon < \mu(0) \\ 0, & \text{if } \epsilon > \mu(0) \end{cases}$

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The Bose Einstein condensation and the black body radiation are left for the near future, whenever we have a chance.

We must move to Ch 6 fast ...